# Synthesis of 2,5-Disubstituted 6-Azaindoles from Substituted Aziridines via Intramolecular Cyclization

## LETTERS 2012 Vol. 14, No. 12 3120–3122

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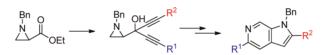
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### Received May 1, 2012

#### ABSTRACT



A new and efficient preparation of pharmacologically and biologically important 2,5-disubstituted 6-azaindoles was achieved from cyclizations of aziridin-2-yl dipropargylic alcohols as adducts of two propargyl groups to ethyl 1-benzylaziridine-2-carboxylate. The sequential cyclizations include pyrrole formation and a novel base-catalyzed intramolecular acetylenic Schmidt reaction.

Azaindole structural units are found in natural marine products, and the importance of their derivatives is reflected in biologically active compounds as indole isosteres.<sup>1</sup> Several strategies for azaindole synthesis have been reported by Reissert, Batcho-Leimgrube, Hemetsberger-Knitel, and Bartoli.<sup>2</sup> However, these methods need improvement in terms of long reaction times, harsh reaction conditions, and low yields. Yet, recently better preparatory methods have come to light such as the Aza-Fischer reaction;<sup>2</sup> transition metal catalyzed cyclizations<sup>4</sup> using Pd, Ru, Zr, Ti, and Cu; and a dilithiation pathway.<sup>5</sup> However, previous synthetic methods cannot provide functionalized 6-azaindoles in high yields.

Compounds containing 6-azaindole show pharmacological activity<sup>1,6</sup> as HIV-1 inhibitors and have a characteristic photophysical property.<sup>7</sup> The synthetic routes for these compounds are limited. Most azaindole syntheses start from substituted pyridines to form a pyrrole unit.<sup>3–5,8</sup> Here, we would like to propose a new efficient synthetic route for the 2,5-substituted 6-azaindoles using an intramolecular pyrrole formation followed by a pyridine ring formation *via* an intramolecular azide cyclization reaction into the acetylenic unit.

We recently reported an efficient synthesis of 1,2,3,5substituted pyrroles from *N*-benzylaziridine-2-carboxylate (1),<sup>9</sup> which is a readily available starting material. The ester 1 was reacted with 2 equiv of various lithium acetylides to

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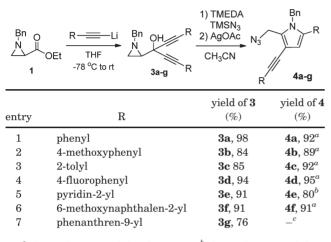
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provide the corresponding dipropargylic alcohols **3** in high yields. This time we were able to prepare 1,2,3,5-substituted pyrroles which have a 2-azidomethyl substituent and various acetylenic substituents at the C-3 position *via* regioselective aziridine ring opening<sup>10</sup> by TMSN<sub>3</sub>, followed by silver ion catalyzed intramolecular cyclization (Scheme 1 and Table 1).

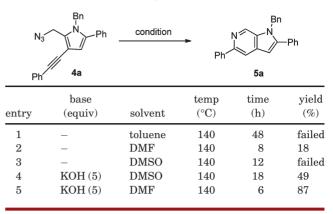
Table 1. Preparation of Substituted Pyrrole 4



<sup>*a*</sup> The reaction was carried out in CH<sub>3</sub>CN. <sup>*b*</sup> The reaction was carried out in DCM. <sup>*c*</sup> The reaction was carried out in DCM/CH<sub>3</sub>CN. Not isolated.

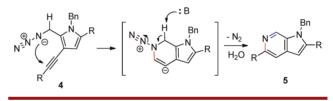
With the 2-azidomethyl-substituted pyrroles, we carried out intramolecular acetylenic Schmidt reactions<sup>11</sup> for the synthesis of azaindoles in various reaction conditions. Unlike usual indolization reactions in the presence of a pyrrole ring, pyrrole-fused pyridine cyclization methods to provide azaindoles are rare<sup>12</sup> and introducing various substituents in the azaindole ring is another challenging task.

In the reactions of modified intramolecular azido, cyclization provides 6-azaindoles in high yield without using metal catalysts (Table 2).<sup>13</sup> When the reaction was carried out without a base in DMSO and toluene, we could not obtain any product (entries 1 and 3). But in DMF we obtained the expected 6-azaindole in 18% yield (entry 2). However, when we added a hydroxide base (KOH), the cyclization reaction proceeded smoothly to provide the corresponding azaindole (entries 4 and 5). After nitrogen cyclization to the acetylenic unit, the deprotonation followed by N<sub>2</sub> elimination seemed to be the driving force to form 2,5-substituted 6-azaindoles in both DMSO (49%) and DMF (87%). Table 2. Formation of 2,5-Diphenyl-6-azaindole



The mechanism of this modified intramolecular azido cyclization of 2-azidomethyl-substituted pyrroles **4** with a hydroxide base to form 6-azaindoles **5** is proposed with the intramolecular base-catalyzed Schmidt cyclization shown in Scheme 1. The anionic nitrogen of the azide attacks the acetylene to form a six-membered ring, and then deprotonation of the benzylic proton by the hydroxide base releases nitrogen gas to promote aromatization. As far as we know, this is the first example of 6-endo intramolecular acetylenic Schmidt cyclization that is free from any metal catalyst.<sup>14</sup>

Scheme 1. Proposed Mechanism of 6-Azaindoles Formation



This synthetic method was used to introduce various substituents on the 2- and 5-positions of 6-azaindoles. The cyclization reactions provided 6-azaindoles with diverse aryl groups bearing electron donors, electron acceptors, and bulky substituents. In all cases, intramolecular cyclization was succeeded to provide 6-azaindoles 5a-5g in high yields of 74–87% (Table 3).

We also prepared 6-azaindoles which have different substituents on the 2- and 5-positions from heterosubstituted bispropargylic alcohols **3h**, **3i**, and **3j** synthesized by two successive alkylation processes using different acetylene reagents.

Low temperature alkynylations provided the corresponding ketones **2** in good yields. Subsequent addition of one more alkyne afforded tertiary bispropargylic alcohols **3h**, **3i**, and **3j** with two different substituents  $R^1$  and  $R^2$  (Table 4).

When pyrrole cyclization was carried out with the tertiary alcohols containing two different acetylenic

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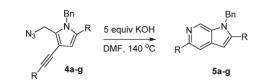
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<sup>(14)</sup> The intramolecular acetylenic Schmidt cyclization was first reported in ref 11, which was a gold(II)-catalyzed 5-exo process.

 Table 3. Preparation of 2,5-Disubstituted 6-Azaindoles (5) from

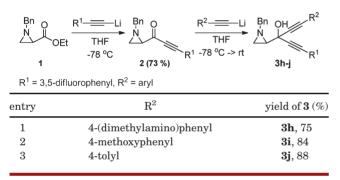
 Pyrroles (4)



entry	R	yield of <b>5</b> (%)
1	phenyl	<b>5a</b> , 87
2	4-methoxyphenyl	<b>5b</b> , 84
3	2-tolyl	<b>5c</b> , 74
4	4-fluorophenyl	<b>5d</b> , 84
5	pyridin-2-yl	<b>5e</b> , 75
6	6-methoxynaphthalen-2yl	<b>5f</b> , 78
7	phenanthren-9-yl	<b>5g</b> , 76 <sup><i>a</i></sup>

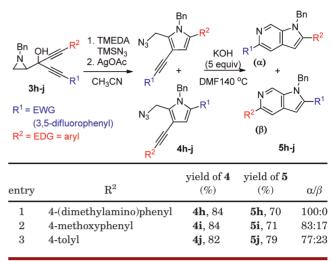
<sup>a</sup> Three-step yield.

Table 4. Preparation of Bispropargylic Alcohols



substituents, the regioselectivity of the cyclization was controlled by the electronic character of the alkynic bond. It was observed that only one pyrrole product 4 h $\alpha$  was obtained when we had a strongly electron-donating 4--(dimethylamino)phenyl substituent and a strongly electron-withdrawing 3.5-difluorophenyl (Table 5, entry 1). However, as the difference between EDG and EWG became smaller, i.e. from 4-(dimethylamino)phenyl to 4-methoxyphenyl and 4-tolyl, the regioselectivity between  $\alpha$  and  $\beta$  in the cyclization became poorer, i.e. 83:17 and 77:23 (Table 5). Regioselectivity of the pyrrole formation was controlled by the electronic character of the triple bond. A silver cation activates the comparably electronrich acetylene between two substituents. As a consequence the pyrrole formation proceeds regioselectively toward the triple bond containing an electron-donating substituent.

The structure of the 6-azaindole **5h** was confirmed by NOE experiments (Figure 1). These data confirm the structure of the 6-azaindole **5h** and also show that intramolecular pyrrole cyclization occurred selectively to more electron-rich acetylenic systems. **Table 5.** Regioselective Formation of Pyrroles 4 and 6-Azain-doles 5 from Aziridin-2-yl Bispropargylic Alcohols 3



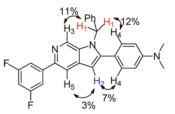
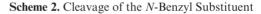
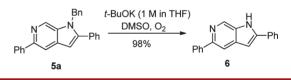


Figure 1. NOE interaction of 6-azaindole 5h.





The *N*-benzyl group originating from aziridine was easily removed with *t*-BuOK to provide 6-azaindole **6**, which has a free pyrrole unit in high yield (Scheme 2).<sup>15</sup>

In conclusion, 2,5-disubstituted 6-azaindoles were efficiently synthesized in high yields by an intramolecular azido cyclization reaction to the acetylenic group of azidomethyl substituted pyrroles which were prepared from substituted aziridines.

Acknowledgment. The authors are grateful for the financial support from the National Research Foundation of Korea (NRF-2010-0005538 and NRF-2009-0081956 for W.K.L. and NRF-2010-0008424 and NRF-2011-0012379 for H.J.H.).

**Supporting Information Available.** Experimental details and copies of <sup>1</sup>H and <sup>13</sup>C spectra for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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The authors declare no competing financial interest.